

# FAST Sn-ION TRANSPORT ON Nb SURFACE FOR GENERATING Nb<sub>x</sub>Sn THIN FILMS AND XPS DEPTH PROFILING\*

Z. Sun<sup>†</sup>, M. Liepe, J. T. Maniscalco, T. Oseroff, and R. D. Porter,  
 CLASSE, Cornell University, Ithaca, NY, USA

D. Zhang, Mechanical Engineering, Cornell University, Ithaca, NY, USA

X. Deng, Chemical Engineering, University of Virginia, Charlottesville, VA, USA

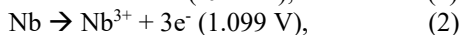
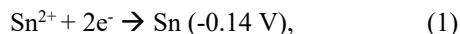
## Abstract

In this work, we propose and demonstrate a fast and facile approach for Nb<sub>x</sub>Sn thin film deposition through the ion exchange reaction. By simply dipping a tin precursor on the Nb substrate surface, a ~600 nm thin film is generated due to the electronegativity difference between Sn and Nb. Through X-ray photoelectron spectroscopy (XPS) depth profiling, the compositional information as a function of film thickness was obtained. Results showed a Sn layer on the film surface, Sn-rich and Nb-rich Nb<sub>x</sub>Sn layers as the majority of the film, and a ~60 nm Nb<sub>3</sub>Sn layer at the film/substrate interface. Quantitative analysis confirmed stoichiometric Nb/Sn ratio for the Nb<sub>3</sub>Sn layer. This deposition method is demonstrated to be an alternative choice for Nb<sub>3</sub>Sn film growth.

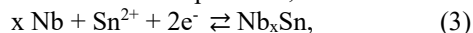
## INTRODUCTION

Nb<sub>3</sub>Sn thin film growth has drawn increased attention recently due to the potential of this material for replacing niobium (Nb) in superconducting radio frequency (SRF) cavities [1, 2]. Nb<sub>3</sub>Sn can lead to superior RF performance owing to its high critical temperature (twice that of Nb) and also a higher superheating field. Considerable efforts have been made to explore Nb<sub>3</sub>Sn film deposition on Nb substrate. The state-of-the-art approach relies on a tin (Sn) vapor diffusion process under >1000 °C high temperature in a high vacuum furnace [3]. Electroplating [4], sputtering [5], and chemical vapour deposition [6] are also actively being developed to coat Nb<sub>3</sub>Sn-based cavities.

In this paper, we report an alternative method for generating Nb<sub>x</sub>Sn films based on fast Sn-ion transport on a Nb surface. Fast ion transport is a fast and facile approach which takes advantage of ion exchange reactions [7, 8]. As illustrated in Fig. 1a, the Nb-Sn alloy can be formed at room temperature when the Sn<sup>2+</sup> cation containing organic electrolyte is exposed to a sufficiently polarized Nb surface. This fast reaction is enabled by the electronegativity difference between Nb and Sn [7]. Accordingly, the half reactions of Sn<sup>2+</sup> reduction and Nb oxidation,



yield a positive electrochemical potential, so the reaction



\*This work was supported by the U.S. National Science Foundation under Award PHY-1549132, the Center for Bright Beams. Also, this work made use of the Cornell Center for Materials Research Shared Facilities which are supported through the NSF MRSEC program (DMR-1120296).

<sup>†</sup> zs253@cornell.edu

is able to proceed spontaneously. Fig. 1b shows the grey-color Nb<sub>x</sub>Sn film from this Sn-ion transport process.

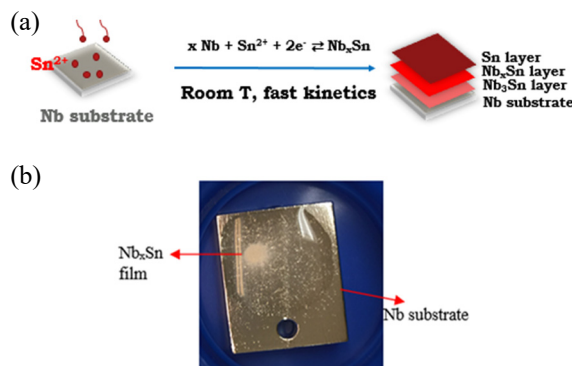


Figure 1: (a) Schematic showing the process of fast Sn-ion transport to the Nb surface. (b) Image of the generated Nb<sub>x</sub>Sn film on the Nb substrate.

In order to further improve the Nb<sub>x</sub>Sn film to a pure-stoichiometric Nb<sub>3</sub>Sn film, the compositional characteristics are of great interest. The use of X-ray photoelectron spectroscopy (XPS) is specialized to characterize the composition at nano-scale (3-5 nm electron mean free path for Sn and Nb elements). By exploiting argon ion sputtering on the film surface and softly etching the film for a few nanometer, the depth profile of the film composition is analyzed in this study.

## EXPERIMENTAL PROCEDURES

The Sn precursor, tin bis(trifluoromethanesulfonyl) imide (SnTFSI), was obtained from Alfa Aesar. 10 mM concentration solution was prepared by dissolving the precursor in DI water. 0.1 mL SnTFSI solution was dropped on the Nb substrate and then dried for 30 min. All these treatments were done in a glove box at room temperature. The film was observed immediately after the drying step.

XPS spectra were collected using a PHI Versaprobe III system with a sensitivity of <0.1% atomic percentage. The system used monochromatic Al k-alpha X-rays with energy of 1486.6 eV and beam size of 200 μm. Ion sputtering was performed for 5.5 min in between each scan for data acquisition. In total, scans 1-35 were taken from the film surface to the film/substrate interface after each sputtering process. The etching rate of 3.6 nm/min was determined from standard SiO<sub>2</sub> films [9]. This work assumes the same sputter rate for Nb<sub>x</sub>Sn films. Thus, each scan probed the film with a step in depth of ~20 nm.

Content from this work may be used under the terms of the CC BY 3.0 licence (© 2019). Any distribution of this work must maintain attribution to the author(s), title of the work, publisher, and DOI

The quantification is based on the integration of Nb 3d and Sn 3d peak areas. The effective atomic counts are affected by the relative sensitivity factor (RSF) between two elements, the electron mean free path (MFP, 1/3 of traveling distance for majority of collected photoelectron), and the transmission correction (T). The effective atomic counts for quantifying the Nb/Sn ratio are calculated by

$$\text{Effective atomic counts} = \frac{\text{Peak area}}{\text{RSF} * T * \text{MFP}} \quad (4)$$

Additionally, the film surface was evaluated with a scanning electron microscope (SEM, Zeiss Gemini 500).

## RESULTS AND DISCUSSION

### Film Surface: Sn Layer

At the surface 80 nm-thin region, as shown in Fig. 2, XPS scans 1-5 observed identical shape and intensity for Nb 3d and Sn 3d peaks. The Sn peak intensity is prominent, while the Nb peak intensity is nearly zero, which suggests the film surface region is a Sn layer.

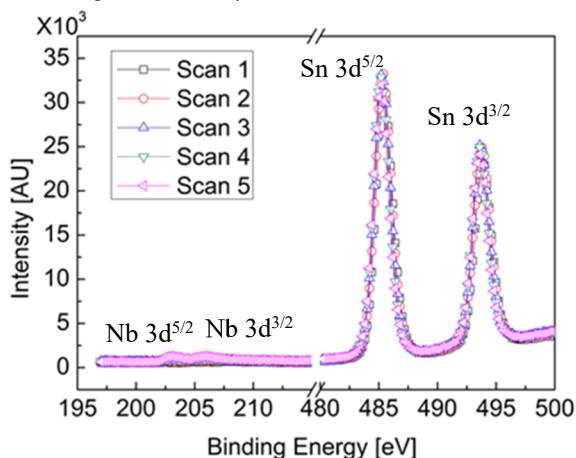


Figure 2: XPS depth profiling of film surface (Scans 1-5) showing a Sn layer.

### Majority of the Film: Nb<sub>x</sub>Sn Layer

After probing the film beyond the 80 nm surface region, a ~420 nm thick Nb<sub>x</sub>Sn layer was detected at scans 6-28. Roughly, this Nb<sub>x</sub>Sn layer was divided into two roughly equal parts — a Sn-rich Nb<sub>x</sub>Sn region (scans 6-18) and a Nb-rich Nb<sub>x</sub>Sn region (scans 20-28).

As shown in Fig. 3a and 3b, the Nb intensity greatly increased along with the decrease of the Sn intensity from scan 6 to 18. However, the Nb/Sn ratio is still below 1.

Starting from scan 20, as shown in Fig. 3c, the Nb intensity became saturated. This saturation indicates that our measurement was not affected by the Nb substrate; otherwise, the Nb intensity would have continued to increase with the film being etched off from the substrate. The Sn intensity (Fig. 3d) continued to decrease, resulting in a Nb-rich Nb<sub>x</sub>Sn region where x is between 1 and 3.

### Film/Substrate Interface: Nb<sub>3</sub>Sn Layer

Further ion sputtering approached the interface between the film and substrate. The identification of this interface can be verified by observation of an increased Nb intensity

after scan 34, where signal from the Nb substrate was significantly detected.

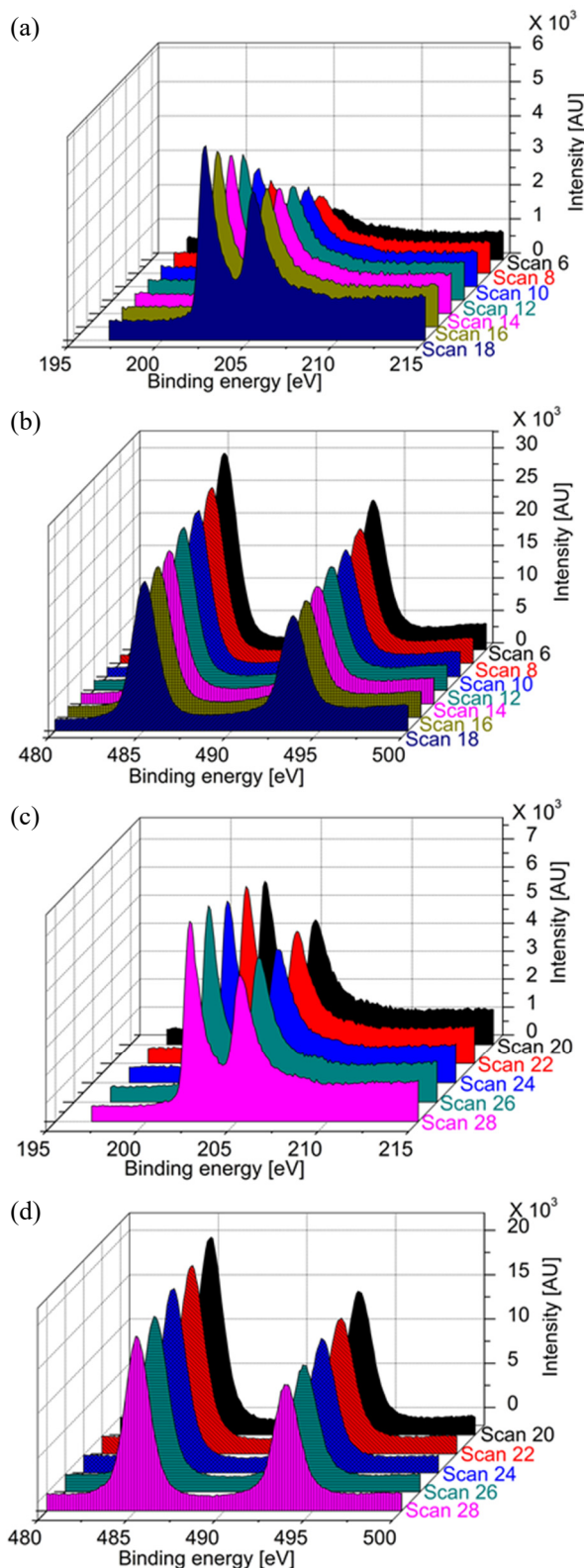


Figure 3: XPS depth profiling of majority of the film (Scans 6-28) showing the Nb<sub>x</sub>Sn layer. Sn-rich Nb<sub>x</sub>Sn layer (Scans 6-18): (a) Nb 3d, (b) Sn 3d. Nb-rich Nb<sub>x</sub>Sn layer (Scans 20-28): (c) Nb 3d, (d) Sn 3d.

Table 1: Calculation of Nb/Sn Atomic Ratio at the Film/Substrate Interface

| Element | Peak area | Relative sensitivity factor <sup>[10]</sup> | Mean free path <sup>[11]</sup> | Transmission correction <sup>[10]</sup> | Effective atomic counts | Atomic ratio |
|---------|-----------|---|--------------------------------|---|-------------------------|--------------|
| Nb      | 26330     | 8.21  | 3                              | 400                                     | 2.67                    | 2.93         |
| Sn      | 49100     | 25.05                                       | 4.9                            | 440                                     | 0.91                    |              |

We now focus on the spectra taken from the interface (scans 29-32). Figure 4 shows that the Nb and the Sn spectra of these scans are identical. More importantly, this 60 nm-thin region is stoichiometric Nb<sub>3</sub>Sn. A detailed calculation of Nb/Sn compositional ratio for scan 31 is summarized in Table 1. The Nb 3d and Sn 3d characteristic peaks were first levelled using a straight background line. Their 3d peak areas were integrated without differentiating orbitals. The corresponding RSF values were obtained from the CasaXPS database. The RSF value for the Sn element is ~3 times higher than Nb, which greatly reduced the effective atomic counts from the Sn component. The resultant 2.93 ratio approximately matches with the composition ratio of Nb<sub>3</sub>Sn.

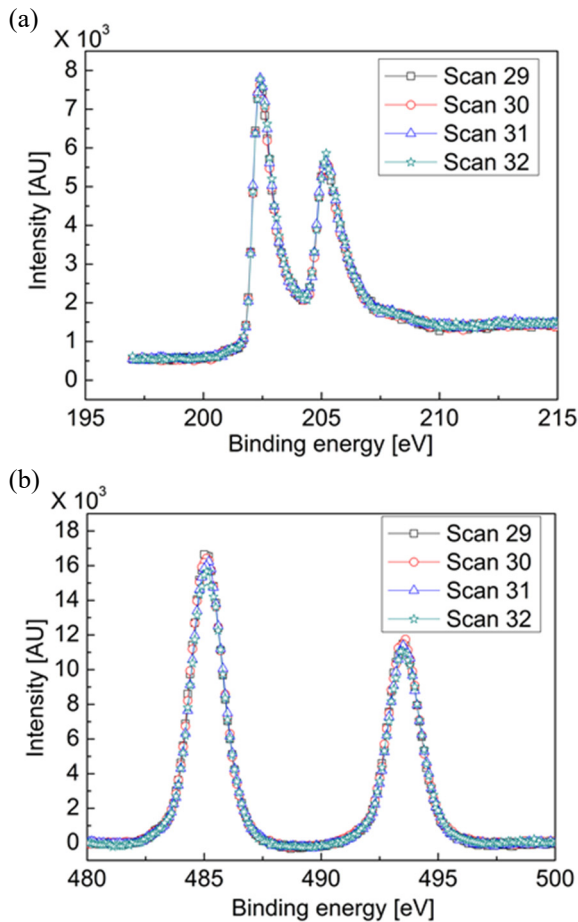


Figure 4: XPS depth profiling of the interface between the film and substrate (Scans 29-32) showing the Nb<sub>3</sub>Sn layer: (a) Nb 3d, (b) Sn 3d.

## Discussion

The 60 nm-thin Nb<sub>3</sub>Sn layer is still smaller than the RF field penetration depth of ~170 nm for Nb<sub>3</sub>Sn [2]. It is necessary to perform a heat treatment to improve the stoichiometry of the Nb<sub>x</sub>Sn layer and increase the thickness of the Nb<sub>3</sub>Sn layer to >200 nm.

Also, the film surface, shown in the SEM image (Fig. 5), is less smooth than our electroplated Sn films [4]. Hence, after achieving required thickness of Nb<sub>3</sub>Sn, a surface smoothing step would be beneficial since surface roughness is believed to be a critical issue impacting RF performance.

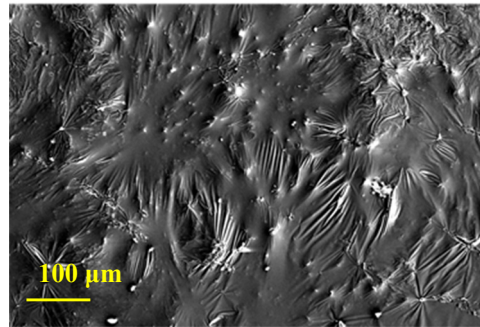


Figure 5: SEM image showing the film surface morphology.

## CONCLUSIONS

In conclusion, a fast Sn-ion transport process was successfully developed to generate Nb<sub>x</sub>Sn films on Nb substrate. XPS depth profile revealed four regions from the film surface to the film/substrate interface: a ~80 nm Sn layer; a ~220 nm Sn-rich Nb<sub>x</sub>Sn layer; a ~200 nm Nb-rich Nb<sub>x</sub>Sn layer; and a ~60 nm Nb<sub>3</sub>Sn layer. The stoichiometry of the Nb<sub>3</sub>Sn layer was accurately quantified, and a Nb/Sn ratio of 2.93 was obtained. Further investigation is required to increase the thickness of the Nb<sub>3</sub>Sn layer and reduce surface roughness.

## REFERENCES

- [1] S. Posen and D.L. Hall, “Nb<sub>3</sub>Sn superconducting radiofrequency cavities: fabrication, results, properties, and prospects”, *Supercond. Sci. Technol.*, vol. 30, p. 033004, 2017.
- [2] S. Posen, M. Liepe, and D.L. Hall, “Proof-of-principle demonstration of Nb<sub>3</sub>Sn superconducting radio frequency cavities for high Q<sub>0</sub> applications”, *Appl. Phys. Lett.*, vol. 106, p. 082601, 2018.
- [3] S. Posen and M. Liepe, “Advances in development of Nb<sub>3</sub>Sn superconducting radio-frequency cavities”, *Phys. Rev. ST Accel. Beams*, vol. 17, p. 112001, 2014.

Content from this work may be used under the terms of the CC BY 3.0 licence (© 2019). Any distribution of this work must maintain attribution to the author(s), title of the work, publisher, and DOI

- [4] Z. Sun *et al.*, “Electroplating of Sn Film on Nb Substrate for Generating Nb<sub>3</sub>Sn Thin Films and Post Laser Annealing”, presented at the 19th Int. Conf. RF Superconductivity (SRF'19), Dresden, Germany, Jun.-Jul. 2019, paper MOP014.
- [5] E.A. Ilyina *et al.*, “Development of sputtered Nb<sub>3</sub>Sn films on copper substrates for superconducting radio frequency applications”, *Supercond. Sci. Technol.*, vol. 32, p. 035002, 2019.
- [6] T. E. Oseroff *et al.*, “Performance of Samples With Novel SRF Materials and Growth Techniques”, in *Proc. 9th Int. Particle Accelerator Conf. (IPAC'18)*, Vancouver, Canada, Apr.-May 2018, pp. 2475-2478. doi:10.18429/JACoW-IPAC2018-WEPMF047
- [7] Z. Tu *et al.*, “Fast ion transport at solid–solid interfaces in hybrid battery anodes”, *Nat. Energy*, vol. 3, p. 310, 2018.
- [8] M. Winter and J. Besenhard, “Electrochemical lithiation of tin and tin-based intermetallics and composites”, *Electrochimica Acta*, vol. 45, p. 31, 1999.
- [9] Z. Sun, X. Deng *et al.*, “Silicon surface passivation by laser processing a sol–gel TiOx thin film”, *ACS Appl. Energy Mater.*, vol. 1, p. 5474, 2018.
- [10] CasaXPS database.
- [11] R. Smart *et al.*, [https://mmrc.caltech.edu/SS\\_XPS/XPS\\_PPT/XPS\\_Slides.pdf](https://mmrc.caltech.edu/SS_XPS/XPS_PPT/XPS_Slides.pdf)